PHOTOCHEMICAL REACTIONS OF PHOSPHAALKENES Abdelkader MERIEM, Jean-Pierre MAJORAL*, Monique REVEL and Jacques NAVECH*

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Summary : Photolysis of phosphaalkenes leads to an inversion of polarity of the phosphorus carbon double bond and to a cleavage into transient phosphinidene and carbene ; unexpected alcoholysis of a P_{III} -C bond and the [4+2] cycloaddition of a linear phosphaalkene on a diene are also reported.

Very few papers have described the photochemical behavior of phosphorus derivatives with a phosphorus carbon double bond. Most of them are concerned with the irradiation of phosphorus ylide $Ph_2P=CRR^{1}$ ¹.

Our interest both in photochemical reactions and in unusually hybridized phosphorus compounds led us to study the photolysis of dicoordinated phosphorus derivatives of general formula R-P=X (X = C< ; N- ; P-). We report here results concerning phosphaalkenes.

Compound <u>1</u>² slowly dimerizes into <u>2</u> at room temperature. However, photolysis of pure <u>2</u> at 254 nm leads quantitatively to monomer <u>1</u> which is transformed on prolonged irradiation (two days) into pentaphenylcyclopentaphosphane <u>3</u>, tetraphenylcyclotetraphosphane <u>4</u> (in the ratio 4:1) and to bis 1,2-(dimethylamino)ethylene. These experimental facts allow us to postulate the transient formation of phenylphosphinidene and dimethylamino carbene. These results are corroborated by the photochemical reaction observed with derivative <u>6</u> (thermally more stable than <u>1</u>). In this case, the transient dimethylaminomethyl carbene gives rise to bis 2,3-(dimethylamino)2-butene and to isopropylene methylamine resulting from migration of a methyl group from the nitrogen to carbon.

When the irradiation of <u>6</u> is performed in benzene solution in the presence of methanol, besides the cyclophosphanes <u>3</u> and <u>4</u>, phenyldimethoxyphosphine <u>8</u>, phenylphosphine <u>9</u> and the methanol adduct <u>7</u> (two diastereoisomers, $\delta^{31}P = +117$ and +118 p.p.m) are obtained. Prolonged photolysis only affords a mixture of <u>8</u> and <u>9</u> (70/30). We showed that <u>8</u> and <u>9</u> came from subsequent irradiation of 3 with methanol ³,⁴.



Note that 7 undergoes an unusual alcoholysis of the P_{III}-C bond giving <u>8</u>.

Surprisingly heating of $\underline{6}$ in acetonitrile methanol solution, in the presence (or not) of base, leads to the formation of the other [1-2] addition product $\underline{10}$ which is in equilibrium with the starting product $\underline{6}$ and with $\underline{9}$.

$$\underline{6} \xrightarrow{\Delta, MeOH}_{-MeOH} Ph-P-C \xrightarrow{NMe_2}_{-Me} \xrightarrow{+MeOH}_{-MeOH} \underline{9} + (MeO)_2 C \xrightarrow{NMe_2}_{-Me}$$

Thus it seems that the polarity of the phosphorus carbon double bond dramatically changes when 6 is thermally or photochemically excited.

Furthermore we observe an "umpolung" effect induced by the dimethylamino group during the thermal addition of methanol to $\underline{6}$. Indeed Bickelhaupt ⁵ described the reverse addition of methanol to 11 in the presence of sodium methoxide, yielding $\underline{12}$.



Only a few cycloaddition reactions of phosphaalkenes have been reported until now ⁶. Most of them involve cycloaddition of cyclic monomers mainly obtained by heating the corresponding dimers or tervalent phospholes. Diels Alder reactions failed with <u>11</u> ⁷ certainly because of steric hindrance. Refluxing <u>1</u> in benzene in the presence of excess of 2,3-dimethyl butadiene affords, besides <u>2</u>, two phosphorus compounds ($\delta^{31}P = -46$, -41 p.p.m). The major one ($\delta^{31}P = -46$ p.p.m, 40%, b.p. 60°/1 mm, mass fragmentation M⁺ 247, 232, 217, 203, 190, 165 ...) has been characterized as <u>13</u>. <u>13</u> is produced in better yield (80%) when <u>1</u> or <u>2</u> are irradiated in acetonitrile or benzene : in this case, all dimer <u>2</u> is transformed into 1 which gives 13 by a thermal cycloaddition ⁸.



During the irradiation of <u>1</u> or <u>2</u> in the presence of 2,3-dimethylbutadiene, neither phospholene <u>14</u>, resulting from a possible reaction of phenylphosphinidene with 2,3-dimethylbutadiene, nor <u>3</u> and <u>4</u>, occuring from the initial formation of phenylphosphinidene or from a photochemical fragmentation of <u>14</u>⁹, were detected in the resulting mixture by ³¹P n.m.r



On the other hand photolysis of <u>6</u> in benzene solution and in presence of 2,3-dimethyl butadiene gives <u>14</u> ($\delta^{31}P = -33$ ppm, yield 15%), <u>15</u> ($\delta^{31}P = -23$ ppm, yield 75%). No direct [4+2] cycloaddition product can be detected in this case.

Moreover the irradiation of <u>6</u> with tolane leads to three phosphorus derivatives. Two of them were identified as pentaphenylcyclopentaphosphane <u>3</u> and 1,2,3,4-tetraphenyl 1,2-diphosphetene <u>16</u> ($\delta^{31}P = -37,5$ ppm; mp : 158-159°; yield 20%). We do not succeed yet in the isolation of the third one ($\delta^{31}P = +3$ ppm). Compound <u>16</u> was already obtained (^{10,11}) by reacting <u>3</u> with tolane in forcing conditions (240 °C, 6 hrs, yield 15%) (¹²).



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- ³ Moreover the fact that photolysis of a non-deoxygenated solution of 3 gives 18 ($\delta^{31}P = +25,2 \text{ p.p.m. J}_{p-H} = 558 \text{ Hz}$) as well as <u>8</u> and <u>9</u>, is in good agreement with the preliminary formation of <u>17</u>.

$$\begin{pmatrix} Ph P \end{pmatrix}_{5} \xrightarrow{h \nu} MeOH \begin{bmatrix} Ph - P, OMe \\ H \end{bmatrix} \longrightarrow Ph P(OMe)_{2} + Ph PH_{2} \\ \underline{3} & \underline{17} & \underline{8} & \underline{9} \\ \hline O_{2} & Ph - P, OMe \\ & H & \underline{18} \\ \end{pmatrix}$$

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- ⁸ Addition of sulphur to 13 in refluxing benzene affords 13' in quantitative yield. 13': $\delta^{31}P = +31 \text{ ppm}$; m.p. = 111-112 °C ; mass fragmentation = M+ 279, 223, 191, 159, 138 ; $\delta^{1}H = 3,12$, 3,17 (s, Me), 3,8 (s, NMe₂).



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- ¹⁰ A. Ecker and U. Schmidt, Chem. Ber., 1973, 106, 1453.
- ¹¹ C. Charrier, J. Guilhem and F. Mathey, J. Org. Chem., 1981, 46, 3.
- ¹² It is interesting to note that in our reaction we never observed the formation of pentaphenyl-1,2,3 triphospholene as Ecker ¹⁰ or Charrier and al.¹¹ did during the reaction of (Ph)₅P with tolane. This could imply either a [2+2] cycloaddition of an instable diphosphene Ph-P=P-Ph with tolane or a [1+2] cycloaddition of a phenylphosphinidene with tolane, formation of the three membered ring 19 then insertion of phenylphosphinidene into the cycle 19 with finally formation of 16.



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